

The hydrogen molecule ion

Evaluation of integrals

This problem is straightforward and not too hard if you:

1. Use atomic units.
2. Use the substitution $\mu = \cos \theta$.
3. Remember that $\sqrt{x^2} = |x|$, not $\sqrt{x^2} = x$.

First the **overlap integral**

$$I(R) = \langle \alpha | \beta \rangle = \langle \eta_g(r_\alpha) | \eta_g(r_\beta) \rangle \quad (1)$$

where

$$\eta_g(r) = \frac{1}{\sqrt{\pi}} e^{-r} \quad \text{and} \quad r_\beta = \sqrt{r_\alpha^2 + R^2 - 2r_\alpha R \cos \theta}.$$

We have

$$\begin{aligned} I(R) &= \int_0^{2\pi} d\phi \int_0^\pi \sin \theta d\theta \int_0^\infty r_\alpha^2 dr_\alpha \frac{1}{\pi} e^{-(r_\alpha + r_\beta)} \\ &= \frac{2\pi}{\pi} \int_0^\infty r_\alpha^2 dr_\alpha \int_{-1}^{+1} d\mu e^{-(r_\alpha + \sqrt{r_\alpha^2 + R^2 - 2r_\alpha R \mu})} \\ &= 2 \int_0^\infty r^2 dr e^{-r} \int_{-1}^{+1} d\mu e^{-\sqrt{r^2 + R^2 - 2rR\mu}}. \end{aligned}$$

The angular integral is

$$\text{A.I.} = \int_{-1}^{+1} d\mu e^{-\sqrt{a-b\mu}} \quad [\text{where } a = r^2 + R^2, b = 2rR]. \quad (2)$$

Use the substitution

$$\begin{aligned} x &= -\sqrt{a-b\mu} \\ dx &= -\frac{1}{2} \frac{-b}{\sqrt{a-b\mu}} d\mu = \frac{1}{2} \frac{b}{-x} d\mu \quad \text{whence} \quad d\mu = -\frac{2x}{b} dx \end{aligned}$$

to find

$$\begin{aligned} \text{A.I.} &= -\frac{2}{b} \int_{\mu=-1}^{+1} dx x e^x \quad [\text{use Dwight 567.1...}] \\ &= -\frac{2}{b} [e^x (x-1)]_{\mu=-1}^{+1} \\ &= \frac{2}{b} \left[e^{-\sqrt{a-b\mu}} (\sqrt{a-b\mu} + 1) \right]_{-1}^{+1} \\ &= \frac{2}{b} \left[e^{-\sqrt{a-b}} (\sqrt{a-b} + 1) - e^{-\sqrt{a+b}} (\sqrt{a+b} + 1) \right]. \end{aligned}$$

Now

$$\begin{aligned}\sqrt{a-b} &= \sqrt{r^2 + R^2 - 2rR} = \sqrt{(r-R)^2} = |r-R| \\ \sqrt{a+b} &= \sqrt{r^2 + R^2 + 2rR} = \sqrt{(r+R)^2} = r+R\end{aligned}$$

so

$$\text{A.I.} = \frac{1}{rR} \left[e^{-|r-R|}(|r-R|+1) - e^{-(r+R)}(r+R+1) \right]. \quad (3)$$

The overlap integral then is

$$\begin{aligned}I(R) &= \frac{2}{R} \int_0^\infty dr r e^{-r} \left[e^{-|r-R|}(|r-R|+1) - e^{-(r+R)}(r+R+1) \right] \\ &= \frac{2}{R} \int_0^\infty dr r e^{-r} \left[e^{-|r-R|}(|r-R|+1) \right] - \frac{2}{R} e^{-R} \int_0^\infty dr r e^{-2r} (r+R+1).\end{aligned}$$

$\underbrace{\hspace{15em}}_{\equiv I_1} \quad \underbrace{\hspace{15em}}_{\equiv I_2}$

The second of these is

$$\begin{aligned}I_2 &= -\frac{2}{R} e^{-R} \left[\int_0^\infty dr r^2 e^{-2r} + (R+1) \int_0^\infty dr r e^{-2r} \right] \quad \text{[[use Dwight 567.2 and 567.1 ...]]} \\ &= -\frac{2}{R} e^{-R} \left[\left[e^{-2r} \left(-\frac{r^2}{2} - \frac{r}{2} - \frac{1}{4} \right) \right]_0^\infty + (R+1) \left[e^{-2r} \left(-\frac{r}{2} - \frac{1}{4} \right) \right]_0^\infty \right] \\ &= -\frac{2}{R} e^{-R} \left[-\left[-\frac{1}{4} \right] + (R+1) \left[\frac{1}{4} \right] \right] \\ &= -e^{-R} \left[\frac{1}{R} + \frac{1}{2} \right].\end{aligned}$$

While the first is

$$\begin{aligned}I_1 &= \frac{2}{R} \int_0^\infty dr r e^{-r} \left[e^{-|r-R|}(|r-R|+1) \right] \\ &= \frac{2}{R} \left[\int_0^R dr r e^{-r} \left[e^{r-R}(-r+R+1) \right] + \int_R^\infty dr r e^{-r} \left[e^{-r+R}(r-R+1) \right] \right] \\ &= \frac{2}{R} \left[e^{-R} \int_0^R dr [-r^2 + (R+1)r] + e^R \int_R^\infty dr e^{-2r} [r^2 - (R-1)r] \right] \\ &= \frac{2}{R} \left[e^{-R} \int_0^R dr [-r^2 + (R+1)r] + e^R \int_R^\infty dr r^2 e^{-2r} - e^R(R-1) \int_R^\infty dr r e^{-2r} \right] \\ &= \frac{2}{R} \left[e^{-R} \left[-\frac{r^3}{3} + (R+1)\frac{r^2}{2} \right]_0^R + e^R \left[e^{-2r} \left(-\frac{r^2}{2} - \frac{r}{2} - \frac{1}{4} \right) \right]_R^\infty - e^R(R-1) \left[e^{-2r} \left(-\frac{r}{2} - \frac{1}{4} \right) \right]_R^\infty \right] \\ &= \frac{2}{R} \left[e^{-R} \left[-\frac{R^3}{3} + (R+1)\frac{R^2}{2} \right] - e^R \left[e^{-2R} \left(-\frac{R^2}{2} - \frac{R}{2} - \frac{1}{4} \right) \right] + e^R(R-1) \left[e^{-2R} \left(-\frac{R}{2} - \frac{1}{4} \right) \right] \right] \\ &= \frac{2}{R} e^{-R} \left[-\frac{1}{3}R^3 + \frac{1}{2}R^3 + \frac{1}{2}R^2 + \frac{1}{2}R^2 + \frac{1}{2}R + \frac{1}{4} - \frac{1}{2}R^2 - \frac{1}{4}R + \frac{1}{2}R + \frac{1}{4} \right] \\ &= e^{-R} \left[\frac{R^2}{3} + R + \frac{3}{2} + \frac{1}{R} \right].\end{aligned}$$

Then

$$I(R) = I_1 + I_2 = e^{-R} \left[\frac{1}{3} R^2 + R + 1 \right]$$

or, in conventional units

$$I(R) = e^{-R/a_0} \left[\frac{1}{3} \left(\frac{R}{a_0} \right)^2 + \frac{R}{a_0} + 1 \right]. \quad (4)$$

Go on to the **direct integral**

$$D(R) = \left\langle \eta_g(r_\alpha) \left| \frac{1}{r_\beta} \right| \eta_g(r_\alpha) \right\rangle \quad (5)$$

Thus

$$\begin{aligned} D(R) &= \int_0^{2\pi} d\phi \int_0^\pi \sin \theta d\theta \int_0^\infty r_\alpha^2 dr_\alpha \frac{1}{\pi} e^{-2r_\alpha} \frac{1}{\sqrt{r_\alpha^2 + R^2 - 2r_\alpha R \cos \theta}} \\ &= \frac{2\pi}{\pi} \int_0^\infty r^2 dr \int_{-1}^{+1} d\mu e^{-2r} \frac{1}{\sqrt{r^2 + R^2 - 2rR\mu}}. \end{aligned}$$

Now, what is

$$\int_{-1}^{+1} d\mu \frac{1}{\sqrt{a - b\mu}}? \quad [\text{Where } a = r^2 + R^2, b = 2rR.]$$

It is

$$\begin{aligned} \int_{-1}^{+1} d\mu \frac{1}{\sqrt{a - b\mu}} &= \left[-\frac{2}{b} \sqrt{a - b\mu} \right]_{-1}^{+1} = -\frac{2}{b} \left[\sqrt{a - b} - \sqrt{a + b} \right] = \frac{2}{b} \left[\sqrt{a + b} - \sqrt{a - b} \right] \\ &= \frac{1}{rR} \left[\sqrt{r^2 + R^2 + 2rR} - \sqrt{r^2 + R^2 - 2rR} \right] \\ &= \frac{1}{rR} \left[\sqrt{(r + R)^2} - \sqrt{(r - R)^2} \right] \quad \llbracket \text{Dangerous curve ahead!} \rrbracket \\ &= \frac{1}{rR} [(r + R) - |r - R|] \\ &= \frac{1}{rR} \begin{cases} (r + R) - (R - r) & \text{for } r < R \\ (r + R) - (r - R) & \text{for } r > R \end{cases} \\ &= \frac{1}{rR} \begin{cases} 2r & \text{for } r < R \\ 2R & \text{for } r > R \end{cases} \\ &= \frac{2}{rR} \min\{r, R\}. \end{aligned}$$

Thus

$$\begin{aligned} D(R) &= \frac{4}{R} \int_0^\infty r^2 dr \frac{1}{r} \min\{r, R\} e^{-2r} \\ &= \frac{4}{R} \int_0^\infty dr r \min\{r, R\} e^{-2r} \\ &= \frac{4}{R} \int_0^R dr r^2 e^{-2r} + \frac{4}{R} \int_R^\infty dr r R e^{-2r} \\ &= \frac{4}{R} \int_0^R dr r^2 e^{-2r} + 4 \int_R^\infty dr r e^{-2r} \quad \llbracket \text{use Dwight 567.2 and 567.1...} \rrbracket \end{aligned}$$

$$\begin{aligned}
&= \frac{4}{R} \left[e^{-2r} \left(\frac{r^2}{-2} - \frac{2r}{4} + \frac{2}{-8} \right) \right]_0^R + 4 \left[e^{-2r} \left(\frac{r}{-2} - \frac{1}{4} \right) \right]_R^\infty \\
&= \frac{4}{R} \left[e^{-2R} \left(\frac{R^2}{-2} - \frac{R}{2} - \frac{1}{4} \right) - \left(-\frac{1}{4} \right) \right] + 4 \left[-e^{-2R} \left(\frac{R}{-2} - \frac{1}{4} \right) \right] \\
&= \frac{1}{R} - \left(1 + \frac{1}{R} \right) e^{-2R}.
\end{aligned}$$

In conventional units,

$$D(R) = \frac{a_0}{R} - \left(1 + \frac{a_0}{R} \right) e^{-2R/a_0}. \quad (6)$$

Onward and upward! Last we work the **exchange integral**

$$\begin{aligned}
X(R) &= \left\langle \eta_g(r_\alpha) \left| \frac{1}{r_\alpha} \right| \eta_g(r_\beta) \right\rangle \quad \text{with } \eta_g(r) = \frac{1}{\sqrt{\pi}} e^{-r} \\
&= \int_0^{2\pi} d\phi \int_0^\pi \sin\theta d\theta \int_0^\infty r_\alpha^2 dr_\alpha \frac{1}{\pi} e^{-(r_\alpha+r_\beta)} \frac{1}{r_\alpha}
\end{aligned} \quad (7)$$

but $r_\alpha + r_\beta = r_\alpha + \sqrt{r_\alpha^2 + R^2 - 2r_\alpha R \cos\theta}$, so

$$\begin{aligned}
X(R) &= \frac{2\pi}{\pi} \int_0^\infty r_\alpha dr_\alpha \int_{-1}^{+1} d\mu e^{-(r_\alpha + \sqrt{r_\alpha^2 + R^2 - 2r_\alpha R \mu})} \\
&= 2 \int_0^\infty r dr e^{-r} \int_{-1}^{+1} d\mu e^{-\sqrt{r^2 + R^2 - 2rR\mu}}
\end{aligned}$$

The angular integral is the same one defined in equation (2) and evaluated in equation (3). It follows that

$$\begin{aligned}
X(R) &= \frac{2}{R} \int_0^\infty dr e^{-r} \left[e^{-|r-R|} (|r-R| + 1) - e^{-(r+R)} (r+R+1) \right] \\
&= \underbrace{\frac{2}{R} \int_0^\infty dr e^{-r} \left[e^{-|r-R|} (|r-R| + 1) \right]}_{\equiv X_1} - \underbrace{\frac{2}{R} e^{-R} \int_0^\infty dr e^{-2r} (r+R+1)}_{\equiv X_2}.
\end{aligned}$$

The second of these is

$$\begin{aligned}
X_2 &= -\frac{2}{R} e^{-R} \left[\int_0^\infty dr r e^{-2r} + (R+1) \int_0^\infty dr e^{-2r} \right] \quad \text{[[use Dwight 565.1 and 567.1...]]} \\
&= -\frac{2}{R} e^{-R} \left[\left[e^{-2r} \left(-\frac{r}{2} - \frac{1}{4} \right) \right]_0^\infty + (R+1) \left[e^{-2r} \left(-\frac{1}{2} \right) \right]_0^\infty \right] \\
&= -\frac{2}{R} e^{-R} \left[-\left[-\frac{1}{4} \right] + (R+1) \left[\frac{1}{2} \right] \right] \\
&= -e^{-R} \left[\frac{3}{2R} + 1 \right].
\end{aligned}$$

While the first is

$$\begin{aligned}
X_1 &= \frac{2}{R} \int_0^\infty dr e^{-r} \left[e^{-|r-R|} (|r-R| + 1) \right] \\
&= \frac{2}{R} \left[\int_0^R dr e^{-r} \left[e^{r-R} (-r + R + 1) \right] + \int_R^\infty dr e^{-r} \left[e^{-r+R} (r - R + 1) \right] \right] \\
&= \frac{2}{R} \left[e^{-R} \int_0^R dr (-r + R + 1) + e^R \int_R^\infty dr e^{-2r} (r - R + 1) \right] \\
&= \frac{2}{R} \left[e^{-R} \int_0^R dr (-r + R + 1) + e^R \int_R^\infty dr r e^{-2r} + e^R (1 - R) \int_R^\infty dr e^{-2r} \right] \\
&= \frac{2}{R} \left[e^{-R} \left[-\frac{r^2}{2} + (R+1)r \right]_0^R + e^R \left[e^{-2r} \left(-\frac{r}{2} - \frac{1}{4} \right) \right]_R^\infty + e^R (1 - R) \left[e^{-2r} \left(-\frac{1}{2} \right) \right]_R^\infty \right] \\
&= \frac{2}{R} \left[e^{-R} \left[-\frac{R^2}{2} + (R+1)R \right] - e^R \left[e^{-2R} \left(-\frac{R}{2} - \frac{1}{4} \right) \right] - e^R (1 - R) \left[e^{-2R} \left(-\frac{1}{2} \right) \right] \right] \\
&= \frac{2}{R} e^{-R} \left[\frac{1}{2} R^2 + R + \frac{1}{2} R + \frac{1}{4} - \frac{1}{2} R + \frac{1}{2} \right] \\
&= e^{-R} \left[R + 2 + \frac{3}{2R} \right].
\end{aligned}$$

Then

$$X(R) = X_1 + X_2 = e^{-R} \left[R + 2 + \frac{3}{2R} - \frac{3}{2R} - 1 \right] = e^{-R} [R + 1].$$

In conventional units,

$$X(R) = e^{-R/a_0} \left[1 + \frac{R}{a_0} \right].$$

I have plotted the integrals $I(R)$, $D(R)$, and $X(R)$ as functions of R using an excel spreadsheet. But I haven't been able to insert the plot into this document because excel (like most Micro\$oft products) seems to be brain dead. You may download the spreadsheet as `HydrogenMoleculeIon.xls`.

Thinking about integrals

The nuclear potential energy is easy: it's

$$\frac{e^2}{4\pi\epsilon_0} \frac{1}{R} \quad \text{or, in atomic units,} \quad \frac{1}{R}$$

The electronic potential and kinetic energies are a bit harder. We will use atomic units throughout. The trial wavefunction is

$$\psi_+(\mathbf{r}) = C_+ [\eta_\alpha(\mathbf{r}) + \eta_\beta(\mathbf{r})] = C_+ (|\alpha\rangle + |\beta\rangle)$$

where $\eta(\mathbf{r}) = e^{-r}/\sqrt{\pi}$.

The mean kinetic energy is

$$\begin{aligned}
\langle \widehat{KE} \rangle &= C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{KE}) (|\alpha \rangle + |\beta \rangle) \right] \\
&= C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{KE}) |\alpha \rangle \right] + C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{KE}) |\beta \rangle \right] \\
&= 2C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{KE}) |\alpha \rangle \right] \\
&= 2C_+^2 \left[\langle \alpha | \widehat{KE} | \alpha \rangle + \langle \beta | \widehat{KE} | \alpha \rangle \right]
\end{aligned}$$

The value of $\langle \alpha | \widehat{KE} | \alpha \rangle$ comes from direct calculation, or from the virial theorem ($\langle \widehat{KE} \rangle = -\frac{1}{2} \langle \widehat{PE} \rangle$ for the hydrogen atom), or from looking it up in a book. The answer is $\langle \widehat{KE} \rangle = +\frac{1}{2}$ (in conventional units, $\langle \widehat{KE} \rangle = +\text{Ry}$).

Meanwhile

$$\begin{aligned}
\widehat{KE} \eta(r) &= -\frac{1}{2} \left[\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial \eta(r)}{\partial r} \right) \right] \\
&= -\frac{1}{2} \eta(r) + \frac{1}{r} \eta(r)
\end{aligned}$$

so

$$\begin{aligned}
\langle \beta | \widehat{KE} | \alpha \rangle &= -\frac{1}{2} \langle \beta | \alpha \rangle + \left\langle \beta \left| \frac{1}{r_\alpha} \right| \alpha \right\rangle \\
&= -\frac{1}{2} I(R) + X(R).
\end{aligned}$$

Recalling that $C_+^2 = 1/[2(1 + I(R))]$ gives

$$\langle \widehat{KE} \rangle = \frac{1}{2} \frac{1 - I(R) + 2X(R)}{1 + I(R)}.$$

We could find the potential energy either directly or else (given that we know $\langle \widehat{H} \rangle$) through $\langle \widehat{H} \rangle = \langle \widehat{KE} \rangle + \langle \widehat{PE} \rangle$. The direct approach is

$$\begin{aligned}
\langle \widehat{PE} \rangle &= C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{PE}) (|\alpha \rangle + |\beta \rangle) \right] \\
&= C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{PE}) |\alpha \rangle \right] + C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{PE}) |\beta \rangle \right] \\
&= 2C_+^2 \left[\langle (\alpha | + \langle \beta |) (\widehat{PE}) |\alpha \rangle \right] \\
&= 2C_+^2 \left[\langle \alpha | \widehat{PE} | \alpha \rangle + \langle \beta | \widehat{PE} | \alpha \rangle \right] \\
&= -2C_+^2 \left[\left\langle \alpha \left| \frac{1}{r_\alpha} + \frac{1}{r_\beta} \right| \alpha \right\rangle + \left\langle \beta \left| \frac{1}{r_\alpha} + \frac{1}{r_\beta} \right| \alpha \right\rangle \right] \\
&= -2C_+^2 \left[\left\langle \alpha \left| \frac{1}{r_\alpha} \right| \alpha \right\rangle + D(R) + 2X(R) \right]
\end{aligned}$$

The value of $\langle \alpha | \widehat{PE} | \alpha \rangle$ comes from direct calculation, or from the virial theorem, or from looking it up in a book. The answer is $\langle \widehat{PE} \rangle = -1$ (in conventional units, $\langle \widehat{PE} \rangle = -2\text{Ry}$).

Recalling that $C_{\pm}^2 = 1/[2(1 + I(R))]$ gives

$$\langle \widehat{PE} \rangle = -\frac{1 + D(R) + 2X(R)}{1 + I(R)}.$$

The values of these energies are calculated and plotted in the excel spreadsheet

`HydrogenMoleculeIon.xls`.

The plot shows energies in atomic units... remember that an energy of $\frac{1}{2}$ in atomic units corresponds to one Ry.